



Energy Conservation and Transformation



Goals

- ✓ Understand how energy can change
- ✓ Observe the transformation of energy
- ✓ Make calculations based on data



Background

Energy is what allows all objects in the universe to move. The energy of atoms or molecules and the energy of stars or galaxies is all the same, just at different sizes. Though we talk about energy being consumed, lost, or used up, it can never be destroyed. It can also never be created. The only thing that energy can do is transform from one kind to another.

Using this fuel cell car, we can use the chemical potential energy of hydrogen to create electrical energy, which will be turned into kinetic energy to cause the car to move. But there are other ways that energy is transformed, even in this small car, which mean that not all the energy in each transformation remains in a useable form.

Thermal energy is an example of a type of energy that isn't always useful. Though we can use it for some applications, such as cooking food, the transformation of different kinds of energy into heat energy is usually a bad thing for most machines. In the case of a car, more heat energy means less kinetic energy, so a smaller percentage of the energy put into the car is used to actually run it.

Fuel cells are much more energy efficient than the internal combustion gasoline engines that power most cars today, but they still have their sources of inefficiency.



Procedure

1. Insert the cylinders into the frame of the car. Fill them with about 40 mL of distilled water.
2. Uncap the tube on the O₂ side of the fuel cell.
3. Fill the syringe with distilled water and fill the fuel cell using the syringe.
4. Replace the cap on the O₂ tube.
5. Insert the fuel cell into the frame of the car in front of the cylinders. Attach the H₂ and O₂ sides of the fuel cell to the H₂ and O₂ cylinders with the longer tubes, which will prevent the hydrogen and oxygen gases from escaping.
6. Connect the battery pack to the fuel cell using the red and black plugs, then turn on the battery pack. You should see the fuel cell start to generate hydrogen and oxygen gas.
7. Once you see bubbles start to escape the H₂ cylinder, turn off and disconnect the battery pack.
8. Connect the red and black wires to the car chassis to start the car.



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Observations



Experimentation

1. You've produced hydrogen and oxygen from water. Now, connect the fuel cell to the motor. What happens?
2. What could you change about your car that might make the car run faster? Try it and observe what happens.
3. What if you wanted to make your car run for a longer time? Would you change the same thing or something different? Try it and observe what happens.



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Measurement

For this section, you will need a multimeter or the Horizon Renewable Energy Monitor. For an introduction to using a multimeter, [click here](#).

1. Measure the current in Amps and the voltage in Volts while generating hydrogen and oxygen. Record your answers below:

Current: _____ A

Voltage: _____ V

2. Voltage is equal to the current multiplied by the resistance ($V = IR$), so according to your data what is the resistance of the fuel cell?

Resistance: _____ Ω

3. Lift the front wheels to keep the car in one place and measure the current in Amps and the voltage in Volts while the car is running. Record your answers below:

Current: _____ A

Voltage: _____ V

4. Why is there a difference between the current/voltage when producing hydrogen and the current/voltage when the car is running? Where has the energy gone?



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Conclusions

1. What kinds of energy did you observe while running your experiments with the fuel cell car?
2. Describe the ways that energy changed from one form to another during this activity.
3. Describe three ways that energy was transformed that didn't help your car move faster or farther.
4. Would it ever be possible to use 100% of the electric energy produced by the fuel cell to move the car? Why or why not?